

CONTENTS

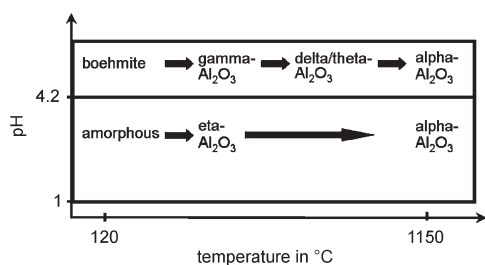
Abstracted/indexed in BioEngineering Abstracts, Chemical Abstracts, Coal Abstracts, Current Contents/Physics, Chemical, & Earth Sciences, Engineering Index, Research Alert, SCISEARCH, Science Abstracts, and Science Citation Index. Also covered in the abstract and citation database SCOPUS[®]. Full text available on ScienceDirect[®].

Regular Articles

Aluminum speciation and thermal evolution of aluminas resulting from modified Yoldas sols

M. Dressler, M. Nofz, F. Malz, J. Pauli, C. Jäger, S. Reinsch and G. Scholz

page 2409



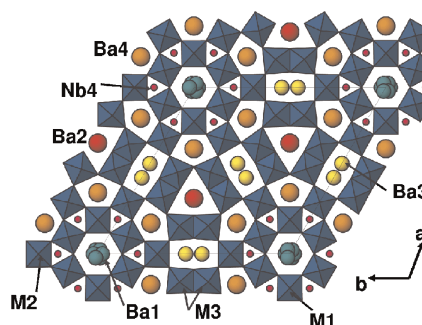
Influence of pH value of parent alumina sols on thermal evolution of resulting xerogels.

Regular Articles—Continued

Contrasting oxide crystal chemistry of Nb and Ta: The structures of the hexagonal bronzes BaTa₂O₆ and Ba_{0.93}Nb_{2.03}O₆

W.Gus Mumme, Ian E. Grey, Robert S. Roth and Terrell A. Vanderah

page 2429

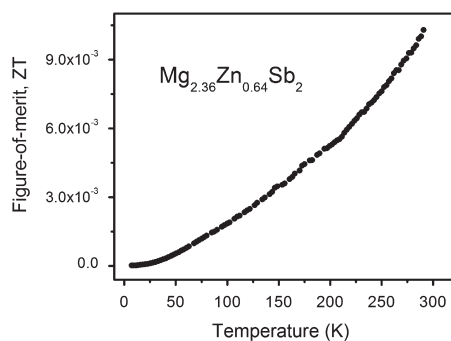


Polyhedral representation of the structure of BaM₂O₆ projected along [001].

Structural and physical properties of Mg_{3-x}Zn_xSb₂ (x = 0–1.34)

Faraz Ahmadpour, Taras Kolodiaznyh and Yuriy Mozharivskij

page 2420

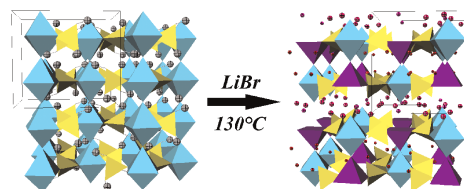


The Mg atoms in Mg₃Sb₂ were successfully substituted with Zn, with Zn going exclusively into the tetrahedral sites. Zn substitution increases the electrical conductivity in Mg_{2.36}Zn_{0.64}Sb₂ by closing the band gap. This change combined with a decrease in the thermal conductivity improves the ZT value.

From a 3D protonic conductor VO(H₂PO₄)₂ to a 2D cationic conductor Li₄VO(PO₄)₂ through lithium exchange

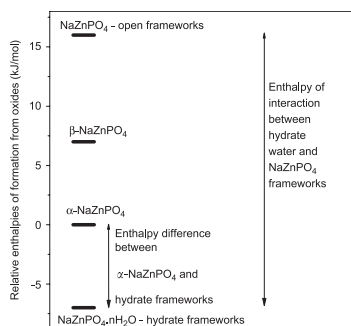
V. Caignaert, M. Satya Kishore, V. Pralong, B. Raveau, N. Creon and H. Fjellvåg

page 2437



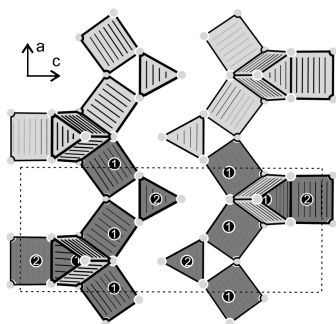
Li₄VO(PO₄)₂ was synthesized by a lithium ion exchange reaction from VO(H₂PO₄)₂. The structure, determined from neutron and synchrotron powder diffraction data, changes in dimensionality from 3D to 2D. VO(H₂PO₄)₂ exhibits a protonic conductivity of 10⁻⁴ S/cm at room temperature while Li₄VO(PO₄)₂ is a lithium conductor.

Energetics of phosphate frameworks containing zinc and cobalt: NaZnPO_4 , $\text{NaH}(\text{ZnPO}_4)_2$, $\text{NaZnPO}_4 \cdot \text{H}_2\text{O}$, $\text{NaZnPO}_4 \cdot \frac{4}{3}\text{H}_2\text{O}$, and $\text{NaCo}_x\text{Zn}_{1-x}\text{PO}_4 \cdot \frac{4}{3}\text{H}_2\text{O}$
So-Nhu Le and Alexandra Navrotsky
page 2443



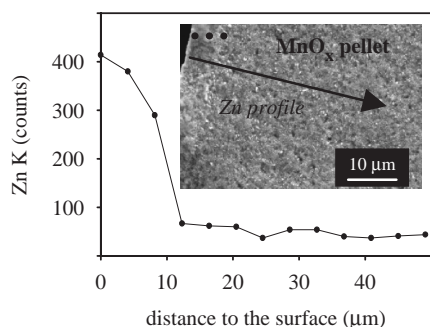
Relative stability of NaZnPO_4 dense phases, open frameworks, and hydrated frameworks. Enthalpy of interaction between water and NaZnPO_4 frameworks is presented by reaction: NaZnPO_4 (cr, open framework) + $n\text{H}_2\text{O}$ (l) \rightarrow $\text{NaZnPO}_4 \cdot n\text{H}_2\text{O}$ (cr, hydrated framework).

Rare earth metal-rich indides $\text{RE}_{14}\text{Rh}_{3-x}\text{In}_3$ ($\text{RE} = \text{Y}, \text{Dy}, \text{Ho}, \text{Er}, \text{Tm}, \text{Lu}$)
Roman Zaremba and Rainer Pöttgen
page 2452



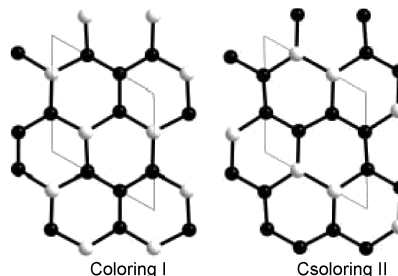
Condensed $[\text{RhRE}_6]$ trigonal prisms in $\text{RE}_{14}\text{Rh}_3\text{In}_3$ indides.

Diffusion and reactivity of $\text{ZnO}-\text{MnO}_x$ system
M. Peiteado, A.C. Caballero and D. Makovec
page 2459



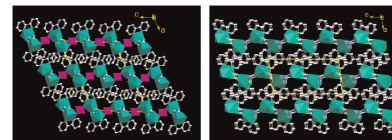
Penetration profile of Zn into MnO_x pellet after firing the ZnO/MnO_x couple at 973 K for 12 h. The formation of a layer of ZnMn_2O_4 spinel phase at the surface of MnO_x pellet will impede further diffusion of zinc and manganese species.

Structural arrangements of the ternary metal boride carbide compounds MB_2C_4 ($M = \text{Mg}, \text{Ca}, \text{La}$ and Ce) from first-principles theory
Chang-Ming Fang, Joseph Bauer, Jean-Yves Saillard and Jean-François Halet
page 2465



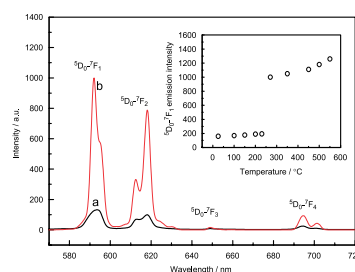
Density-functional theory calculations on the structural arrangements of the ternary metal borocarbides MB_2C_4 ($M = \text{Mg}, \text{Ca}; \text{La}$ and Ce) indicate that these compounds adopt a layered structure consisting of graphite-like B_2C_4 layers alternating with metal sheets. Within the hexagonal layers, the coloring with the $-\text{C}-\text{C}-\text{B}-\text{C}-\text{B}-$ sequence is energetically more stable than that with the $-\text{C}-\text{C}-\text{C}-\text{B}-\text{B}-$ one.

Syntheses, crystal structures and photoluminescence properties of three novel organically bonded indium selenates or selenites
Mei-Ling Feng, Heng-Yun Ye and Jiang-Gao Mao
page 2471



Three new organically bonded indium(III) selenates or selenites hybrids have been synthesized and characterized. Their structures feature organically bonded indium oxalate selenate or selenite layer or hydrogen-bonded 3D structures based on indium selenate dimers.

Thermal and photoluminescence properties of hydrated $\text{YPO}_4:\text{Eu}^{3+}$ nanowires
Weihua Di, Xiaoxia Zhao, Shaozhe Lu, Xiaojun Wang and Haifeng Zhao
page 2478

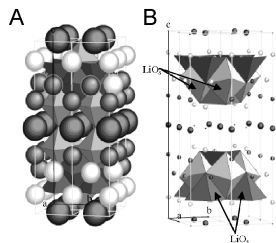


The emission efficiency of the sample annealed at 250 °C is about six times as high as that of as-synthesized sample. This is attributed primarily to the coordinated water as the nonradiative relaxation pathways in the as-synthesized sample.

Continued

Synthesis and CO₂ capture evaluation of Li_{2-x}K_xZrO₃ solid solutions and crystal structure of a new lithium-potassium zirconate phase

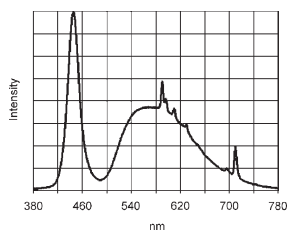
Mayra Y. Veliz-Enriquez, Gonzalo Gonzalez and Heriberto Pfeiffer
page 2485



Li_{2-x}K_xZrO₃ samples were synthesized and characterized by XRD, SEM, and TGA. The solubility limits of potassium into Li₂ZrO₃ is $x=0.2$. Additionally, a new phase was synthesized, Li_{2.27}K_{1.19}Zr_{2.16}O_{6.05}, which was studied by Rietveld. On the other hand, Li_{2-x}K_xZrO₃ solid solutions were tested as CO₂ captors. Analyses showed that Li_{2-x}K_xZr₂O₃ samples present a better CO₂ absorption than Li₂ZrO₃ pure.

Quantum efficiency of double activated Tb₃Al₅O₁₂:Ce³⁺, Eu³⁺

Mihail Nazarov, Do Young Noh, Jongrak Sohn and Chulsoo Yoon
page 2493

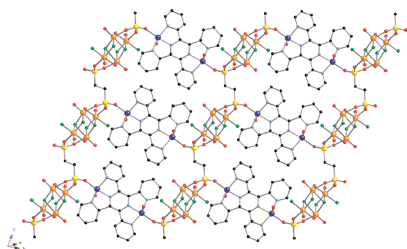


Emission spectra of the blue LED including TAG:Ce, Eu.

Solid state coordination chemistry of the oxofluorovanadium-diphosphonate system in the presence of Cu(II)-tetrapyriddy pyrazine complex cations

The crystal structures of [Cu₂(tpyprz)(H₂O)₂]₄V₄F₆O₆(O₃PCH₂CH₂PO₃)₂, [Cu₂(tpyprz)(H₂O)₂]₄V₄F₆O₆(O₃PCH₂CH₂PO₃)₃, and [Cu₂(tpyprz){HO₃P(CH₂)₃PO₃H}][V₂F₂O₅] (tpyprz = tetra-4-pyridylpyrazine)

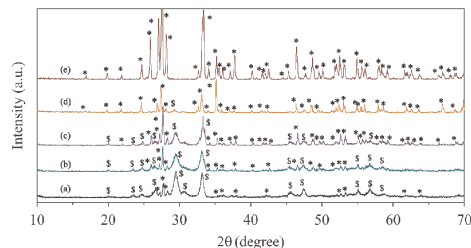
Wayne Ouellette, Vladimir Golub, Charles J. O'Connor and Jon Zubieta
page 2500



Hydrothermal reactions have been exploited in the preparation of a series of bimetallic oxides constructed from {Cu₂(tpyprz)}⁴⁺ and oxyfluorovanadium organophosphonate components. A view of the two-dimensional structure of the mixed valence [Cu₂(tpyprz)(H₂O)₂]₄V₄F₆O₆(O₃PCH₂CH₂PO₃)₂ is provided.

Controlled synthesis of bismuth oxo nanoscale crystals (BiOCl, Bi₁₂O₁₇Cl₂, α-Bi₂O₃, and (BiO)₂CO₃) by solution-phase methods

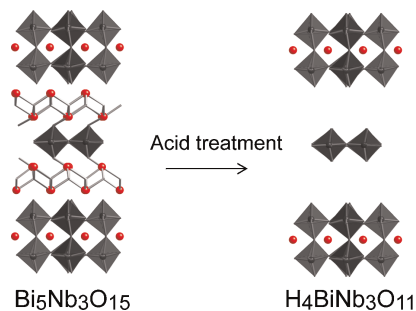
Xiang Ying Chen, Hyun Sue Huh and Soon W. Lee
page 2510



We prepared bismuth oxo nanomaterials by adjusting growth parameters. BiOCl, Bi₁₂O₁₇Cl₂, and α-Bi₂O₃ could be prepared from BiCl₃ and NaOH, whereas (BiO)₂CO₃ was prepared from BiCl₃ and urea. BiOCl and Bi₁₂O₁₇Cl₂ could also be prepared from BiCl₃ and ammonia. The α-Bi₂O₃ sample exhibited strong emission at room temperature.

Characterization of Bi₅Nb₃O₁₅ by refinement of neutron diffraction pattern, acid treatment and reaction of the acid-treated product with n-alkylamines

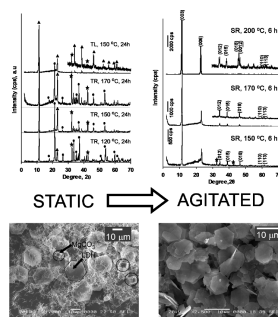
Seichi Tahara, Akira Shimada, Nobuhiro Kumada and Yoshiyuki Sugahara
page 2517



Crystal structure of Bi₅Nb₃O₁₅ is investigated by refinement of neutron diffraction pattern as well as by structural change through acid treatment of Bi₅Nb₃O₁₅ and subsequent treatment of an acid-treated product with *n*-alkylamine.

Short-time hydrothermal synthesis and delamination of ion exchangeable Mg/Ga layered double hydroxides

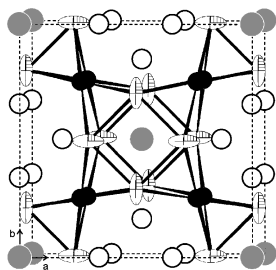
Ugur Unal
page 2525



Hydrothermal synthesis under agitation resulted in highly crystalline Mg/Ga LDHs slabs in a short time. The LDHs slabs were delaminated into two-dimensional nanosize sheets.

Copper–indium ordering in $RECu_6In_6$ ($RE = Y, Ce, Pr, Nd, Gd, Tb, Dy$)

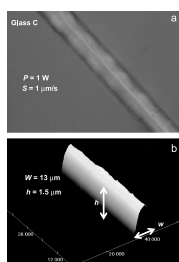
Roman Zaremba, Ihor Muts, Rolf-Dieter Hoffmann, Yaroslav M. Kalychak, Vasyly I. Zaremba and Rainer Pöttgen
page 2534



The copper–indium network in $CeCu_6In_6$.

Line patterning of $(Sr,Ba)Nb_2O_6$ crystals in borate glasses by transition metal atom heat processing

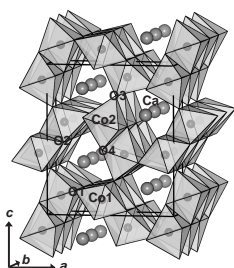
M. Sato, T. Honma, Y. Benino and T. Komatsu
page 2541



This figure shows the polarization optical (a) and confocal scanning laser (b) micrographs for the sample obtained by heat-assisted (300 °C) Nd:YAG laser irradiation with a laser power of $P=1\text{ W}$ and laser scanning speed of $S=1\text{ }\mu\text{m/s}$ in Glass C. The figure demonstrates that the transition metal atom heat processing (i.e., a combination of cw Nd:YAG laser and Ni^{2+} ions) is a novel technique for spatially selected crystallization of SBN crystals in glass.

Structure and properties of the $CaFe_2O_4$ -type cobalt oxide $CaCo_2O_4$

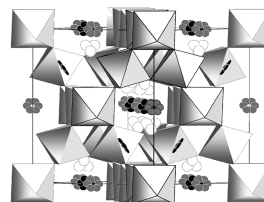
Mitsuyuki Shizuya, Masaaki Isobe and Eiji Takayama-Muromachi
page 2550



A new calcium cobalt oxide $CaCo_2O_4$ phase, which crystallizes in the calcium-ferrite-type structure (space group: $Pnma$; $a=8.789(2)\text{ }\text{\AA}$, $b=2.9006(7)\text{ }\text{\AA}$, and $c=10.282(3)\text{ }\text{\AA}$), has been synthesized for the first time under high temperature and high pressure (1500 °C, 6 GPa). This compound exhibits large thermoelectric power (Seebeck coefficient: $S=+147\text{ }\mu\text{V/K}$ at 380 K) and an unsaturated temperature dependence of S around 380 K.

The disordered structures and low temperature dielectric relaxation properties of two misplaced-displacive cubic pyrochlores found in the $Bi_2O_3-M^{II}O-Nb_2O_5$ ($M = Mg, Ni$) systems

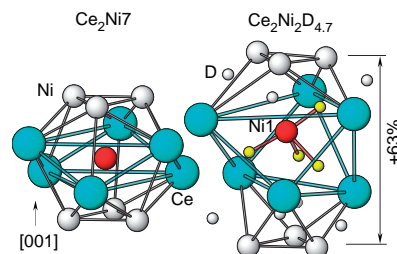
Hai Binh Nguyen, Lasse Norén, Yun Liu, Ray L. Withers, Xiaoyong Wei and Margaret M. Elcombe
page 2558



The final displacively disordered average structure of $Bi_{1.667}Mg_{0.70}Nb_{1.52}O_7$ (BMN) projected along a close to $\langle 110 \rangle$ direction.

Crystal chemistry and thermodynamic properties of anisotropic $Ce_2Ni_7H_{4.7}$ hydride

R.V. Denys, V.A. Yartys, Masashi Sato, A.B. Riabov and R.G. Delaplane
page 2566



An extremely large volume expansion takes place during the hydrogenation of the hexagonal Ce_2Ni_7 intermetallic compound (left). The expansion proceeds exclusively along one crystallographic direction ([001]) and dramatically modifies the metal sublattice. Hydrogen atoms in $Ce_2Ni_7D_{4.7}$ deuteride form an ordered sublattice and preferably fill sites centred by $NiCe_6$ octahedra (right).

Self-catalyst growth of single-crystalline CaB_6 nanostructures

Junqi Xu, Yanming Zhao, Chunyun Zou and Qiwei Ding
page 2577

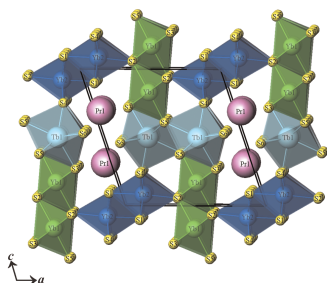


Large-scale calcium hexaboride (CaB_6) nanostructures have been successfully fabricated with self-catalyst method using calcium (Ca) powders and boron trichloride (BCl_3) gas mixed with hydrogen and argon. Our results show that the nanowires are highly single crystals elongated preferentially in the $[1\ 1\ 0]$ direction.

Continued

Syntheses, structure, magnetism, and optical properties of the partially ordered quaternary interlanthanide sulfides $\text{PrLnYb}_2\text{S}_6$ ($\text{Ln} = \text{Tb}, \text{Dy}$)

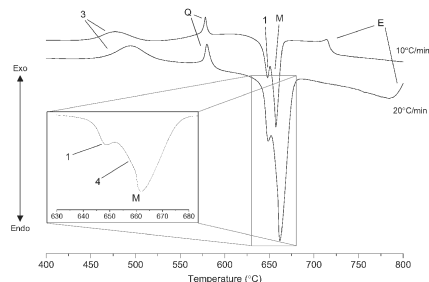
Geng Bang Jin, Eun Sang Choi, Robert P. Guertin, James S. Brooks, Corwin H. Booth and Thomas E. Albrecht-Schmitt
page 2581



An illustration of the three-dimensional structure of $\text{PrTbYb}_2\text{S}_6$ viewed along the b -axis.

Thermal phase transitions in antimony (III) oxides

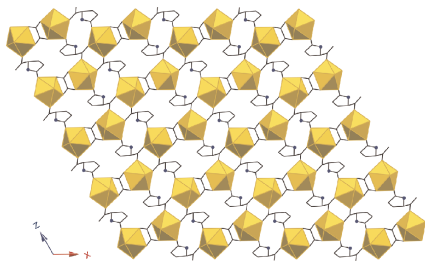
R.G. Orman and D. Holland
page 2587



The phase transitions of Sb_2O_3 have been re-examined using XRD and simultaneous TG/DTA, clarifying apparent disagreements in previous works. The thermal events have been detailed, and the α - β phase transition has been observed to occur as a multi-stage event—possibly relating to the presence of surface- or bulk-bound water.

In situ ligand synthesis with the UO_2^{2+} cation under hydrothermal conditions

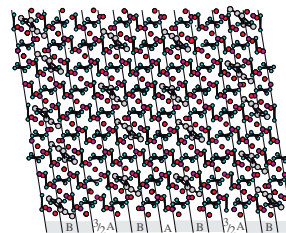
Mark Frisch and Christopher L. Cahill
page 2597



A novel homometallic coordination polymer $(\text{UO}_2)_2(\text{C}_2\text{O}_4)(\text{C}_5\text{H}_6\text{NO}_3)_2$, in the uranium-L-pyroglutamic acid system has been synthesized under hydrothermal conditions. The title compound consists of uranium pentagonal bipyramids bridged through both L-pyroglutamate and oxalate linkages to produce a 3D crystal structure. The oxalate anions are theorized to result from decarboxylation of L-pyroglutamic acid followed by subsequent coupling of CO_2 .

Effect of metal doping on the low-temperature structural behavior of thermoelectric $\beta\text{-Zn}_4\text{Sb}_3$

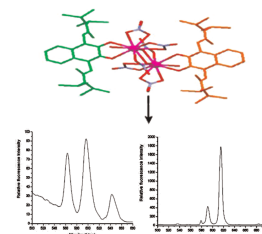
Johanna Nylén, Sven Lidin, Magnus Andersson, Hongxue Liu, Nate Newman and Ulrich Häussermann
page 2603



The thermoelectric material Zn_4Sb_3 displays complex temperature polymorphism. Room temperature stable, disordered, $\beta\text{-Zn}_4\text{Sb}_3$ undergoes two phase transitions at 254 and 235 K to the consecutively higher ordered phases α and α' , respectively. The α - α' transformation is triggered by a slight and homogenous Zn deficiency and introduces a compositional modulation in the $\alpha\text{-Zn}_4\text{Sb}_3$ structure.

Preparation, spectroscopic properties of 1,4-di (*N,N*-diisopropylacetamido)-2,3(1*H*,4*H*)-quinoxalinedione (L) lanthanide complexes and the supramolecular structure of $[\text{Nd}_2\text{L}_2(\text{NO}_3)_6(\text{H}_2\text{O})_2] \cdot \text{H}_2\text{O}$

Xue-Qin Song, Yang Yu, Wei-Sheng Liu, Wei Dou, Jiang-Rong Zheng and Jun-Na Yao
page 2616

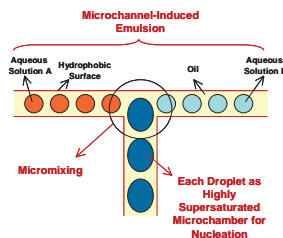


Preparation, spectroscopic properties of 1,4-di (*N,N*-diisopropylacetamido)-2,3(1*H*,4*H*)-quinoxalinedione (L) lanthanide complexes and the supramolecular structure of $[\text{Nd}_2\text{L}_2(\text{NO}_3)_6(\text{H}_2\text{O})_2] \cdot \text{H}_2\text{O}$.

Rapid Communications

Continuous nanoparticle production by microfluidic-based emulsion, mixing and crystallization

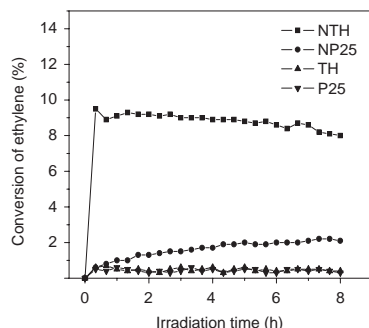
Y.-F. Su, H. Kim, S. Kovenklioglu and W.Y. Lee
page 2625



BaSO_4 and 2,2'-dipyridylamine (DPA) nanoparticles were synthesized as reactive crystallization and anti-solvent recrystallization examples, respectively, of using the microfluidic-based emulsion and mixing approach as a new avenue of continuously producing inorganic and organic nanoparticles.

New synthesis of excellent visible-light $\text{TiO}_{2-x}\text{N}_x$ photocatalyst using a very simple method

Danzhen Li, Hanjie Huang, Xu Chen, Zhixin Chen, Wenjuan Li, Dong Ye and Xianzhi Fu
page 2630

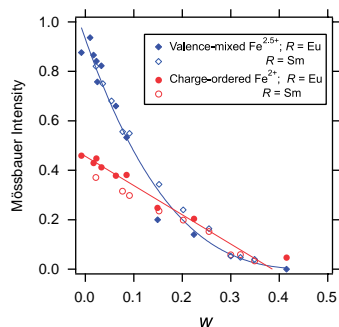


The excellent visible-light-responsive (from 400 to 550 nm) $\text{TiO}_{2-x}\text{N}_x$ photocatalyst was prepared by a simple wet method. Hydrazine was used as a new nitrogen resource in this paper. In the experiment, a strong photocatalytic activity with high photochemical stability under visible-light irradiation was demonstrated.

Corrigendum

Corrigendum to “ $\text{EuBaFe}_2\text{O}_{5+n}$: Valence mixing and charge ordering are two separate cooperative phenomena” [J. Solid State Chem. 180 (2007) 148–157]

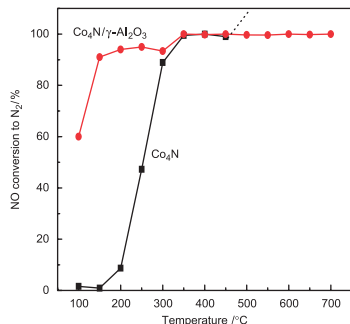
P. Karen, K. Gustafsson and J. Lindén
page 2641



Fraction of $\text{Fe}^{2.5+}$ in the valence-mixed state (curve) compared with the fraction of Fe^{2+} in the charge-ordered state (line).

Synthesis, characterization and activity of alumina-supported cobalt nitride for NO decomposition

Zhiwei Yao, Aimin Zhu, Jing Chen, Xinkui Wang, C.T. Au and Chuan Shi
page 2635



The supported cobalt nitride performs much better than its bulk counterpart for NO decomposition, owing to its small crystal size, high thermal stability and big surface area.

Author inquiries

Submissions

For detailed instructions on the preparation of electronic artwork, consult the journal home page at <http://authors.elsevier.com>.

Other inquiries

Visit the journal home page (<http://authors.elsevier.com>) for the facility to track accepted articles and set up e-mail alerts to inform you of when an article's status has changed. The journal home page also provides detailed artwork guidelines, copyright information, frequently asked questions and more.

Contact details for questions arising after acceptance of an article, especially those relating to proofs, are provided after registration of an article for publication.

Language Polishing

Authors who require information about language editing and copyediting services pre- and post-submission should visit <http://www.elsevier.com/wps/find/authorshome.authors/languagepolishing> or contact authorsupport@elsevier.com for more information. Please note Elsevier neither endorses nor takes responsibility for any products, goods, or services offered by outside vendors through our services or in any advertising. For more information please refer to our Terms & Conditions at http://www.elsevier.com/wps/find/termsconditions.cws_home/termsconditions.

For a full and complete Guide for Authors, please refer to *J. Solid State Chem.*, Vol. 180, Issue 1, pp. *bmi–bmv*. The instructions can also be found at http://www.elsevier.com/wps/find/journaldescription.cws_home/622898/authorinstructions.

Journal of Solid State Chemistry has no page charges.